# Paper 1

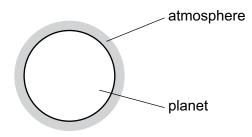
### Questions are applicable for both core and extended candidates

- **1** Some information about gas X is listed.
  - It is not present in clean, dry air.
  - It is not a cause of respiratory problems.
  - It is responsible for global warming.

### What is X?

- A carbon dioxide
- B carbon monoxide
- C methane
- D nitrogen dioxide
- 2 What are the **main** products obtained by the fractional distillation of liquid air?
  - A carbon dioxide and oxygen
  - B carbon dioxide and water vapour
  - **C** nitrogen and oxygen
  - D nitrogen and water vapour
- 3 In which reaction is the rate of reaction increased by light?
  - A carbon dioxide + water  $\rightarrow$  glucose + oxygen
  - B ethanoic acid + sodium carbonate  $\rightarrow$  sodium ethanoate + water + carbon dioxide
  - $\textbf{C} \quad \text{ethene + bromine} \rightarrow \text{dibromoethane}$
  - **D** methane + oxygen  $\rightarrow$  carbon dioxide + water

- 4 Which substance is beneficial to aquatic life?
  - A dissolved oxygen
  - **B** phosphates
  - **C** plastics
  - D sewage
- 5 A new planet has been discovered and its atmosphere has been analysed.



The table shows the composition of its atmosphere.

gas	percentage by volume		
carbon dioxide	4		
nitrogen	72		
oxygen	24		

Which gases are present in the atmosphere of the planet in a higher percentage than they are in the Earth's atmosphere?

- A carbon dioxide and oxygen
- B carbon dioxide only
- **C** nitrogen and oxygen
- D nitrogen only
- 6 Which statement is correct?
  - **A** Atmospheric carbon dioxide is not a cause of climate change.
  - **B** Atmospheric methane is produced by respiration.
  - **C** Burning natural gas decreases the level of carbon dioxide in the atmosphere.
  - **D** Decomposition of vegetation causes an increase in atmospheric methane.

7 A plastic combusts to form sulfur dioxide, SO<sub>2</sub>, and hydrogen chloride, HC*l*.

How could both gases be removed from the air?

- **A** Pass the gases over solid anhydrous cobalt(II) chloride.
- **B** Pass the gases over solid damp calcium oxide.
- **C** Pass the gases through a catalytic converter.
- **D** Pass the gases through filter paper.
- 8 Which information about carbon dioxide and methane is correct?

		carbon dioxide	methane	
A	formed when vegetation decomposes	$\checkmark$	X	key
в	greenhouse gas	1	$\checkmark$	✓ = correct
С	present in unpolluted air	x	X	<b>x</b> = not correct
D	produced during respiration	X	$\checkmark$	

- 9 Which process produces methane?
  - **A** combustion of hydrocarbons
  - **B** decomposition of vegetation
  - **C** respiration
  - D reaction between hydrochloric acid and calcium carbonate
- **10** Which substance in polluted air damages stonework and kills trees?
  - A carbon dioxide
  - B carbon monoxide
  - C lead compounds
  - D sulfur dioxide
- **11** What is the approximate volume of nitrogen in 200 cm<sup>3</sup> of air?

**A**  $20 \text{ cm}^3$  **B**  $40 \text{ cm}^3$  **C**  $80 \text{ cm}^3$  **D**  $160 \text{ cm}^3$ 

## Paper 2

### Questions are applicable for both core and extended candidates unless indicated in the question

- **12** How do carbon dioxide and methane cause global warming? (extended only)
  - **A** They emit the thermal energy they have absorbed back to the Earth.
  - **B** They absorb the radiation directly from the Sun.
  - **C** They increase thermal energy loss to space.
  - **D** They reduce reflection of thermal energy from the Earth's surface.
- **13** Some combustion reactions produce pollutant gases.

Which reactions produce a pollutant gas that is not present in clean air?

- $1 \quad 2CH_4 \ + \ 3O_2 \ \rightarrow \ 2CO \ + \ 4H_2O$
- $2 \quad 2H_2 \ \textbf{+} \ \textbf{O}_2 \ \rightarrow \ \textbf{2}H_2\textbf{O}$
- $3 \quad C + O_2 \rightarrow CO_2$
- $4 \quad N_2 \ + \ O_2 \ \rightarrow \ 2NO$
- **A** 1 and 3 **B** 1 and 4 **C** 2 and 3 **D** 3 and 4

**14** What is the equation for photosynthesis? (extended only)

- $\textbf{A} \quad \text{CO}_2 \ \textbf{+} \ \ \textbf{3H}_2 \ \rightarrow \ \textbf{CH}_3\textbf{OH} \ \textbf{+} \ \ \textbf{H}_2\textbf{O}$
- $\textbf{B} \quad 6CO_2 \ \textbf{+} \ 6H_2O \ \rightarrow \ C_6H_{12}O_6 \ \textbf{+} \ 6O_2$
- $\textbf{C} \quad C_6H_{12}O_6 \ \rightarrow \ 2C_2H_5OH \ + \ 2CO_2$
- $\label{eq:constraint} \begin{array}{ccc} \textbf{D} & C_6H_{12}O_6 \mbox{ + } 6O_2 \mbox{ + } 6CO_2 \mbox{ + } 6H_2O \end{array}$
- **15** Which statement describes how the C–H bonds in methane gas in the atmosphere contribute to global warming? (extended only)
  - **A** They absorb thermal energy from the Sun and emit some of this energy into space.
  - **B** They absorb thermal energy from the Sun and emit all of this energy towards the Earth.
  - **C** They absorb thermal energy from the Earth and emit all of this energy towards the Earth.
  - **D** They absorb thermal energy from the Earth and emit some of this energy towards the Earth.

**16** Catalytic converters in car exhausts change polluting gases into non-polluting gases.

Which statements about oxides of nitrogen and car engines are correct? (extended only)

- 1 The nitrogen in oxides of nitrogen comes from compounds in gasoline.
- 2 The oxygen in oxides of nitrogen comes from the air in the car engine.
- 3 Catalytic converters convert oxides of nitrogen into nitrogen.
- **A** 1 and 2 **B** 2 and 3 **C** 2 only **D** 3 only
- **17** Which statement is correct?
  - **A** Atmospheric carbon dioxide is not a cause of climate change.
  - **B** Atmospheric methane is produced by respiration.
  - **C** Burning natural gas decreases the level of carbon dioxide in the atmosphere.
  - **D** Decomposition of vegetation causes an increase in atmospheric methane.